Chuktabularins E-T, 16-Norphragmalin Limonoids from *Chukrasia tabularis* var. velutina

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Received November 17, 2009

Chuktabularins E–T (1–16), 16 new 16-norphragmalin limonoids, together with four known compounds, chuktabularins A–D, were isolated from the stem bark of *Chukrasia tabularis* var. *velutina*. These compounds possess a biosynthetically extended propionyl or acetyl group at C-15 and a characteristic ketal moiety between the limonoid skeleton and the acyl substituent at C-15. The structures of these compounds were established on the basis of detailed spectroscopic analysis, and that of 1 was confirmed by a single-crystal X-ray diffraction experiment, representing the first verification of the skeleton of 16-norphragmalin limonoids. Chuktabularins K–O (7–11) were found to be the first 19-acetoxylated 16-norphragmalin limonoids. Variable-temperature ¹H NMR experiments suggested that 7 exists as an equilibrium mixture of conformational isomers in solution. The absolute configuration of 5 was determined by the CD exciton chirality method on its 11,12-di-*p*-chlorobenzoate (5a), and those of 1–4 and 6–16 were proposed by correlating with 5 spectroscopically and biogenetically.

Limonoids, distributed mainly in the plant family Meliaceae, have been a focus in natural products research for their structural diversity and potential biological significance.^{1,2} Plants of the genus Chukrasia are well-known for their structurally diversified phragmalintype limonoids,^{1,3} which are B,D-seco limonoids with a characteristic A- and B-ring system of tricyclo[3.3.1^{2,10}.1^{1,4}]decane or tricyclo[4.2.1^{10,30}.1^{1,4}]decane,^{2b,4} and found also in the other Meliaceae tribes, Swietenieae5 and Carapeae.6 Previous chemical studies on this genus have focused mainly on C. tabularis A. Juss., leading to the isolation of several phragmalin limonoids with interesting carbon skeletons,^{2b,7} some of which exhibit insect antifeedant7c and potassium channel-blocking7f activities. Limonoids with a 16-norphragmalin skeleton, isolated from C. tabularis for the first time in 2007, possess an unprecedented ketal moiety between the phragmalin skeleton and a biosynthetically extended acyl substituent at C-15.7d Up to the present time, only eight 16-norphragmalin limonoids have been reported.7d,g,i

Chukrasia tabularis var. velutina, a timber tree, is a variety of C. tabularis and grows mainly in tropical areas of Asia such as India and southern mainland China.8 Its stem bark has been used traditionally as an astringent, antidiarrheal, and anti-influenza agent in the People's Republic of China.^{8b} As a part of our ongoing research program to isolate novel compounds from plants of the Meliaceae, a new class of C-15-acyl phragmalin limonoids featuring a C-16/C-30 δ -lactone ring^{7h} and three 16-norphragmalin limonoids with an unprecedented skeleton⁷ⁱ were isolated from the stem bark of C. tabularis var. velutina. Further work on phragmalin-type limonoids present in the air-dried stem bark afforded 16 new 16norphragmalin limonoids, chuktabularins E-T (1–16), together with four known compounds, chuktabularins A-D.^{7d} These new compounds possess a characteristic ketal moiety between the limonoid skeleton and the biosynthetically extended propionyl or acetyl substituent at C-15.^{7d} The structures of compounds 1-16were established on the basis of detailed spectroscopic analysis, and that of 1 was confirmed by single-crystal X-ray diffraction. Herein, we describe the isolation, structural elucidation, and the plausible biosynthetic origin (Supporting Information, S1) of these new compounds.

Results and Discussion

Chuktabularin E (1) was isolated as colorless crystals, and its molecular formula was established as $C_{36}H_{44}O_{16}$ by the HRESIMS



ion at m/z 755.2527 [M + Na]⁺. The IR absorption bands at 3418 and 1755 cm⁻¹ suggested the presence of hydroxy and ester groups. The ¹H and ¹³C NMR data (Tables 1 and 2) and information from 2D NMR studies (HMBC, HSQC, and NOESY) suggested 1 as being a phragmalin-type limonoid⁷ and revealed the presence of a β -substituted furanyl ring [$\delta_{\rm H}$ 6.43, 7.39, 7.50; $\delta_{\rm C}$ 122.2, 109.5, 143.5, 140.5], an ethyl group [$\delta_{\rm H}$ 2.01 (m, 2H), 1.10 (t, J = 7.6Hz, 3H); $\delta_{\rm C}$ 26.1, 8.2], two angular methyls [$\delta_{\rm H}$ 0.93 (s, 3H), 0.92 (s, 3H); $\delta_{\rm C}$ 18.7, 15.1], and four acetyls. A pair of germinal doublets at $\delta_{\rm H}$ 1.96 (d, J = 11.6 Hz) and 2.06 (d, J = 11.6 Hz) was assigned to the H-29 protons of the characteristic phragmalin 4,29,1-ring bridge, which was confirmed by the HMBC correlations observed from the H-29 protons to the quaternary carbons at $\delta_{\rm C}$ 86.0 (C-1), 85.4 (C-3), 44.7 (C-4), 41.1 (C-5), and 52.3 (C-10) (Figure 1). Two geminal oxygenated methylene signals at $\delta_{\rm H}$ 4.18 and 5.00 (each d, J = 12.6 Hz) corresponded to a ¹³C NMR signal at $\delta_{\rm C}$ 69.4 (C-19), which showed correlations with carbons at $\delta_{\rm C}$ 41.1 (C-5), 75.3 (C-9), and 52.3 (C-10) and suggested that the 19-methyl group is oxygenated. The HMBC correlations between C-7 ($\delta_{\rm C}$ 172.9) and H-6 ($\delta_{\rm H}$ 2.31) and one of the oxygenated C-19 methylene signals at $\delta_{\rm H}$ 5.00 (H-19a) indicated the presence of the characteristic C-6-C-7 appendage of a phragmalin-type limonoid and the C-7/C-19 δ-lactone ring.^{7d,h,9}

A HMBC correlation between H-15a [$\delta_{\rm H}$ 2.52 (dd, J = 12.0, 8.0 Hz)] and the ketal resonance at $\delta_{\rm C}$ 113.7 (C-1'), instead of the correlation between H-15 and the C-16 carbonyl in common phragmalins, indicated that **1** is a 16-decarboxylated phragmalin limonoid. The HMBC correlations between the ketal carbon and the ethyl group signals [$\delta_{\rm H}$ 2.01 (m, 2H) and $\delta_{\rm H}$ 1.10 (t, J = 7.6 Hz, 3H)] suggested the linkage of the ethyl to the ketal carbon, i.e., a biosynthetically extended propionyl group attached at C-15.^{7d}

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Table 1. ¹ H NMR Data of Compounds $1-6$ in C	CDC	1
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position	1 ^{<i>a</i>}	2 ^b	3 ^b	4 ^a	5 ^{<i>a</i>}	6 ^b
3	5.27, s	5.35, s	5.35, s	3.85, s	5.33, s	5.24, s
5	2.07, m ^c	2.08, m ^c	2.06, m ^c	2.32, m ^c	2.88, brdd (9.0, 8.3)	2.91, brdd (9.0, 8.5)
6a	2.31, m, 2H ^c	2.32, m, 2H ^c	2.30, m, 2H ^c	2.35, m, 2H ^c	2.23, dd (14.7, 10.9)	2.48, dd (15.0, 7.2)
6b					2.47, m ^c	2.23, brd (15.0)
11	5.61, d (3.5)	5.64, d (3.6)	5.64, d (3.6)	5.84, d (3.4)	4.20, brs	4.14, brs
12	5.47, d, (3.5)	5.50, d (3.6)	5.52, d (3.6)	5.24, d (3.4)	4.20, brs	4.24, brs
14	3.23, dd (12.0, 8.0)	3.25, dd (12.0, 7.8)	3.26, dd (12.0, 7.8)	3.15, dd (11.6, 8.6)	2.95, dd (11.6, 7.9)	2.93, dd (12.0, 8.4)
15a	2.52, dd (12.0, 8.0)	2.56, dd (12.0, 7.8)	2.56, dd (12.0, 7.8)	2.38, m ^c	2.46, m ^c	2.40, dd (12.0, 8.4)
15b	1.95, m ^c	1.90, dd (12.0, 12.0)	1.91, dd (12.0, 12.0)	1.92, dd (12.0, 11.7)	1.86, dd (11.7, 11.6)	1.92, dd (12.0, 12.0)
17	6.17, s	6.08, s	6.08, s	6.37, s	6.01, s	6.09, s
18	0.93, s, 3H	0.90, s, 3H	0.90, s, 3H	1.01, s, 3H	1.09, s, 3H	1.11, s, 3H
19a	5.00, d (12.6)	5.03, d (12.6)	5.01, d (12.6)	4.98, d (12.7)	4.90, d (12.7)	4.89, d (12.0)
19b	4.18, d (12.6)	4.18, d (12.6)	4.17, d (12.6)	4.16, d (12.7)	4.15, d (12.7)	4.15, d (12.0)
21	7.50, brs	7.48, brs	7.48, brs	7.61, brs	7.30, brs	7.34, brs
22	6.43, brd (1.0)	6.41, brd (0.6)	6.41, brd (0.6)	6.42, brd (1.0)	6.29, brd (1.0)	6.31, brd (0.6)
23	7.39, t-like (1.7)	7.39, t-like (1.2)	7.39, t-like (1.5)	7.37, t-like (1.7)	7.36, t-like (1.7)	7.37, t-like (1.8)
28	0.92, s, 3H	0.90, s, 3H	0.90, s, 3H	1.02, s, 3H	0.92, s, 3H	0.93, s, 3H
29a	2.06, d (11.6)	2.05, m ^c	2.08, d (11.4)	1.98, d (11.5)	2.09, m ^c	2.04, m ^c
29b	1.96, d (11.6)	2.00, m ^c	2.04, d (11.4)	1.91, d (11.5)	1.97, m ^c	2.00, m ^c
30	4.19, s	4.63, s	4.64, s	4.23, s	4.62, s	4.21, s
2'	2.01, m, 2H ^c	2.00, q (7.5), 2H	2.00, q (7.5), 2H	2.00, m, 2H ^c	1.94, m, 2H ^c	2.02, q (7.8), 2H
3'	1.10, t (7.6), 3H	1.07, t (7.5), 3H	1.07, t (7.5), 3H	1.09, t (7.5), 3H	1.05, t (7.5), 3H	1.07, t (7.8), 3H
OH-1	4.45, brs	4.89, brs	4.89, brs	4.15, brs	4.89, brs	4.45, brs
OH-9		3.27, brs	3.22, brs			4.74, brs
OH-11						5.78, brs
OAc-2		2.11, s, 3H	2.10, s, 3H		2.08, s, 3H	
OAc-3	2.40, s, 3H	2.50, s, 3H	2.49, s, 3H		2.34, s, 3H	2.27, s, 3H
OAc-11	2.12, s, 3H	2.11, s, 3H	2.10, s, 3H	2.06, s, 3H		
OAc-12	2.05, s, 3H			2.03, s, 3H		
OCOCH ₂ CH ₃ -12		2.35, q (7.5), 2H 1.10, t (7.50), 3H				
OCOCH(CH ₃) ₂ -12			2.56, m ^c 1.17, d (7.2), 3H 1.15, d (7.2), 3H			
OAc-17	2.12, s, 3H	2.08, s, 3H	2.09, s, 3H	2.12, s, 3H	2.07, s, 3H	2.11, s, 3H

^a Recorded at 500 MHz (¹H) and 125 MHz (¹³C). ^b Recorded at 600 MHz (¹H) and 150 MHz (¹³C). ^c Signal pattern unclear due to overlapping.

The HMBC correlation between H-30 ($\delta_{\rm H}$ 4.19, s) and the ketal carbon was consistent with the presence of an ether bridge between C-1' and C-30. The remaining one degree of unsaturation implied that an additional ring is required. The downfield shifted signal of C-8 ($\delta_{\rm C}$ 89.8) suggested its possible ether linkage to C-1'.^{7d} Accordingly, compound **1** was assigned with a 16-norphragmalin limonoid framework.^{7d} However, this deduction could not be verified directly by the HMBC spectrum since both C-8 and C-1' are quaternary carbons. A single-crystal X-ray diffraction experiment (Figure 2) was performed and confirmed the ether linkage between C-8 and C-1', forming a characteristic 2,7-dioxabicyclo[2.2.1]heptane moiety between the limonoid skeleton and the propionyl group at C-15. Thus, the structure of **1** was determined as shown.

Strong NOESY cross-peaks from H-11 to H-5, H-12, H-17, and H-30, from H-17 to H-12 and H-30, and from H-5 to Me-28 indicated a β -orientation of these six proton signals in **1**. NOESY correlations of H-14 with Me-18, H-29a with H-3, and H-29b with H-19b revealed that these protons adopt an α -orientation. The α -orientation of H-3 was also confirmed by the NOESY correlation between the OAc-3 signal and H-21 of the furan ring. Thus, the relative configuration of **1** in solution was established by a NOESY experiment as depicted (Figure 1), which matched that in the solid state as obtained by the X-ray crystallographic study (Figure 2).

Chuktabularins F (2) and G (3) exhibited quasimolecular ions at m/z 811.2794 [M + Na]⁺ and 825.2938 [M + Na]⁺ in the HRESIMS, consistent with the molecular formulas of C₃₉H₄₈O₁₇ and C₄₀H₅₀O₁₇, respectively. The general features of the ¹H and ¹³C NMR spectroscopic data of **2** and **3** closely resembled those of **1**, with the exception of signals corresponding to a propionyl group [$\delta_{\rm H}$ 2.35 (q, J = 7.5, 2H), 1.10 (t, J = 7.5 Hz, 3H); $\delta_{\rm C}$ 173.5, 27.2, 9.0] and an isobutyryl group [$\delta_{\rm H}$ 2.56 (m), 1.17 (d, J = 7.2 Hz, 3H), 1.15 (d, J = 7.2 Hz, 3H); $\delta_{\rm C}$ 175.9, 34.0, 19.0, 18.8], respectively. Differences were observed in the downfield shifted signals of C-2 ($\Delta \delta_C 7.8$ and 7.9) and H-30 ($\Delta \delta_H 0.44$ and 0.45) in comparison to **1**, which suggested the substitution of the acyl group at C-2 in **2** and **3**.^{7h} Specification of the sites of ester linkages was obtained by HMBC experiments, with three acetyl groups assigned to C-3, C-11, and C-17 and the additional propionyl and isobutyryl group located at C-12 of **2** and **3**, respectively. Thus, the structures of **2** and **3** were established as the 12-*O*-propionyl and 12-*O*-isobutyryl derivatives of 2-*O*-acetylated **1**, respectively. Their relative configurations were established using their NOESY spectra to be the same as that of **1**.

Chuktabularin H (4), a white, amorphous powder, gave the molecular formula $C_{34}H_{42}O_{15}$, as determined by HRESIMS, which showed a quasimolecular ion at m/z 713.2412 [M + Na]⁺. The similarity of the ¹H and ¹³C NMR spectroscopic data of 4 (Tables 1 and 2) to those of 1 indicated that 4 possesses the same 16-norphragmalin skeleton with a characteristic ketal moiety, which was confirmed by the presence of a ketal resonance at δ_C 113.0 and key HMBC correlations. The major difference found was the absence of an acetyl signal at C-3, and as a result, the H-3 signal of 4 was shifted upfield to δ_H 3.85 when compared with 1 (δ_H 5.27), consistent with the lack of any HMBC correlation between H-3 and an acetyl carbonyl carbon. Thus, the structure of 4 was demonstrated to be the 3-*O*-deacetyl derivative of 1.

Chuktabularin I (5), an isomer of 4, gave the same molecular formula of $C_{34}H_{42}O_{15}$, as determined by the HRESIMS ion at m/z713.2419 [M + Na]⁺. The ¹H and ¹³C NMR spectroscopic data of 5 (Tables 1 and 2) indicated that this compound possesses the same 16-norphragmalin skeleton as 4 with a ketal moiety between the main skeleton and the biosynthetically extended propionyl group at C-15 and a C-7/C-19 δ -lactone, and the difference was in the location of the acetyl group. The chemical shift of C-2 was shifted from δ_C 73.4 in 1 to δ_C 81.8 in 5, which suggested that an acetyl

position	1^{a}	2^b	3 ^b	4 ^{<i>a</i>}	5 ^{<i>a</i>}	6 ^b
1	86.0	85.5	85.5	86.1	85.4	85.7
2	73.4	81.2	81.3	73.3	81.8	73.6
3	85.4	82.9	82.9	86.4	83.4	86.0
4	44.7	45.6	45.6	44.6	45.3	44.2
5	41.1	40.5	40.5	39.9	41.3	41.8
6	31.5	31.4	31.4	31.5	31.3	31.9
7	172.9	172.7	172.5	173.3	175.1	175.5
8	89.8	89.6	89.5	89.3	90.4	90.6
9	75.3	75.2	75.2	75.1	76.7	76.7
10	52.3	52.4	52.4	52.4	51.8	51.8
11	71.4	71.3	71.4	71.4	68.6	68.4
12	72.1	71.8	71.6	72.4	75.7	75.7
13	41.6	41.5	41.4	41.5	42.2	42.2
14	43.8	43.9	44.0	43.9	43.0	42.9
15	33.2	34.0	34.0	33.6	34.0	33.4
17	71.1	71.3	71.3	70.9	71.4	71.0
18	18.7	19.1	19.2	18.0	18.9	18.9
19	69.4	69.4	69.3	69.4	69.6	69.7
20	122.2	122.4	122.4	121.2	122.8	122.3
21	140.5	140.2	140.2	141.9	139.8	140.1
22	109.5	109.5	109.5	109.7	109.9	109.8
23	143.5	143.3	143.3	143.2	143.1	143.2
28	15.1	15.0	15.1	15.0	15.0	15.0
29	38.4	38.8	38.8	38.1	39.1	38.6
30	71.8	70.7	70.7	71.8	70.7	71.5
1'	113.7	113.3	113.3	113.0	113.0	113.4
2'	26.1	25.9	25.9	26.0	26.0	26.1
3'	8.2	7.9	7.9	8.0	7.8	8.2
OAc-2		169.5	169.5		169.5	
		20.9	20.8		20.8	
OAc-3	169.6	169.1	169.1		168.9	169.6
	21.1	21.0	21.0		21.4	21.6
OAc-11	171.1	170.9	170.9	170.1		
	20.9	20.9	20.8	20.9		
OAc-12	170.0			169.8		
	20.6			20.6		
OCOCH ₂ CH ₃ -12		173.5				
		27.2				
		9.0				
OCOCH(CH ₃) ₂ -12			175.9			
			34.0			
			19.0			
			18.8			
OAc-17	169.0	168.8	168.8	170.4	168.9	169.1
	20.8	20.5	20.5	21.2	20.5	20.9

^a Recorded at 500 MHz (¹H) and 125 MHz (¹³C). ^b Recorded at 600 MHz (¹H) and 150 MHz (¹³C).



Figure 1. Key HMBC (\rightarrow) and NOE (\leftrightarrow) correlations of **1**.

group is positioned at C-2.^{7h} On the basis of the corresponding HMBC correlations, the other two acetyls were assigned at C-3 and C-17. Thus, the structure of **5** was established as the 2-O-acetyl, 11,12-O-deacetyl derivative of **1**.

The absolute configuration of **5** was determined by applying the CD exciton chirality method on its 11,12-di-*p*-chlorobenzoate (**5a**), for which the CD spectrum exhibited negative chirality resulting

from the exciton coupling between the two chromophores of *p*-chlorobenzoate at 255 ($\Delta \varepsilon - 4.30$, $\pi \rightarrow \pi^*$ transition) and 232 nm ($\Delta \varepsilon + 1.32$, $\pi \rightarrow \pi^*$ transition). The negative chirality indicated that the transition dipole moments of two chromophores are oriented in a counterclockwise manner¹⁰ (Figure 3), corresponding to the absolute configuration of **5a** as 11*R* and 12*R*. Thus, the absolute configuration of **5** was assigned as depicted.



Figure 2. X-ray crystallographic structure of 1.

Chuktabularin J (6) was isolated as a white, amorphous powder, and its molecular formula was established as $C_{32}H_{40}O_{14}$ from the HRESIMS ion at m/z 671.2331 [M + Na]⁺. The similarity of the ¹H and ¹³C NMR spectroscopic data (Tables 1 and 2) of **6** and **5** when compared indicated that these two natural products both possess the same 16-norphragmalin skeleton. However, the C-2 (δ_C 81.8) and H-30 (δ_H 4.62, s) signals of **5** were shifted to δ_C 73.6 and δ_H 4.21 for **6**, which suggested the absence of a 2-*O*-acetyl group in **5**.^{7h} Two acetyls were placed at C-3 and C-17 on the basis of the corresponding HMBC correlations. Thus, the structure of **6** was established as the 2-*O*-deacetyl derivative of **5**.

Chuktabularin K (7) was obtained as a white, amorphous powder. The HRESIMS showed a quasimolecular ion at m/z 857.2841 [M $(+ Na)^+$, consistent with a molecular formula of $C_{40}H_{50}O_{19}$. The IR absorption bands at 3436 and 1751 cm⁻¹ suggested the presence of hydroxy and ester groups. The ¹H and ¹³C NMR data and the information from the subsequent 2D NMR studies (HMBC, HSQC, and NOESY) indicated that 7 is a 16-norphragmalin-type limonoid with a characteristic ketal moiety between the limonoid skeleton and the biosynthetically extended acetyl group at C-15, which was confirmed by the HMBC cross-peaks (Figure 4) from the ketal resonance at $\delta_{\rm C}$ 110.5 (C-1') to H-15a ($\delta_{\rm H}$ 2.55, dd, J = 11.4, 7.8 Hz), a methyl ($\delta_{\rm H}$ 1.63, s, 3H), and H-30 ($\delta_{\rm H}$ 4.64, s). Two proton signals at $\delta_{\rm H}$ 4.46 and 4.64 (brd, J = 11.7 Hz) were assignable to an oxygenated C-19 methylene with the broadened ¹³C NMR signal at $\delta_{\rm C}$ 66–67 from the observed HSQC correlations. The HMBC correlations between C-7 ($\delta_{\rm C}$ 173.1) and H-5 ($\delta_{\rm H}$ 2.70, brd, J =12.0 Hz) and the methoxy protons at $\delta_{\rm H}$ 3.63 (s, 3H) confirmed the characteristic C-6-C-7 appendage of the phragmalin limonoid and also the cleavage of the C-7/C-19 δ -lactone ring. Furthermore, four acetyl groups were assigned at C-3 ($\delta_{\rm C}$ 83.0), C-11 ($\delta_{\rm C}$ 72.0), C-12 ($\delta_{\rm C}$ 72.8), and C-17 ($\delta_{\rm C}$ 71.2) according to their corresponding HMBC correlations. On the basis of the chemical shifts of C-2 ($\delta_{\rm C}$ 82.0) and H-30 ($\delta_{\rm H}$ 4.64, s), the remaining acetyl was attached at C-2.⁷ Two hydroxy proton signals at $\delta_{\rm H}$ 3.55 (s, OH-9) and 4.82 (s, OH-1) were distinguished from the HSQC and HMBC spectra. This evidence permitted the last acetyl to be placed at C-19, the only remaining substitution site. A NOESY experiment of 7 indicated that all the asymmetric carbons have the same relative configuration as those of 1. Strong cross-peaks from H-11 to H-5, H-17, and H-30 and from H-17 to H-12 and H-30 indicated a β -orientation of these protons. NOESY correlations of H-14 with Me-18 and OH-9, and of H-3 with H-29b, revealed that these protons adopt an α -orientation. Thus, the structure of 7 was determined as depicted.

The C-19 methylene protons of **7** in CDCl₃ exhibited two moderately broadened signals at room temperature (303 K) as compared with relatively sharp signals in **1**. Such a phenomenon

was attributed to an equilibrium among various conformations in the molecules at room temperature, resulting from restricted rotation about the C-10, C-19 bond. In an attempt to verify this hypothesis, the ¹H NMR spectrum of **7** was recorded at variable temperatures (303, 318, 323, and 333 K) in DMSO- d_6 .¹¹ Expansions of the spectroscopic regions between δ_H 4.30 and δ_H 4.50 are shown in Figure 5. An increase in temperature caused no splitting but rather a sharpening of the C-19 methylene signals, which suggested that signal coalescence of the conformers occurs below room temperature. These results thus confirmed the exchange broadening of the C-19 methylene resonances due to rotational hindrance at the C-10, C-19 bond.

Chuktabularin L (8) showed a molecular formula of $C_{39}H_{50}O_{18}$ as determined by the HRESIMS ion at m/z 829.2891 [M + Na]⁺. Analysis of its ¹H and ¹³C NMR data (Table 3) indicated that compound 8 is a congener of 7, which possesses a 16-norphragmalin skeleton with an acetoxylated C-19 angular methyl. In the HMBC spectrum of 8, the observed correlations of the ketal resonance at $\delta_{\rm C}$ 113.2 (C-1') with H-15a ($\delta_{\rm H}$ 2.51, dd, J = 12.0, 8.4 Hz), the ethyl signals [$\delta_{\rm H}$ 2.04 (q, J = 7.8 Hz, 2H) and $\delta_{\rm H}$ 1.09 (t, J = 7.8, 3H)], and H-30 ($\delta_{\rm H}$ 4.21, s) suggested that 8 has a propional group at C-15, like **1**. The upfield shifted signal of C-2 ($\delta_{\rm C}$ 73.9) indicated the absence of the acetyl at C-2 of 7. Two hydroxy proton signals at $\delta_{\rm H}$ 3.53 (s, OH-9) and 4.42 (s, OH-1) were distinguished using the HSQC and HMBC spectra. Corresponding HMBC correlations ascertained that the hydroxy groups, at C-3, -11, -12, -17, and -19, are all acetylated. The relative configuration was determined to be the same as that of 7 from the NOESY spectrum. Thus, the structure of 8 was demonstrated as shown.

Chuktabularin L (9), a white, amorphous powder, gave an ion at m/z 871.3001 [M + Na]⁺ in the HRESIMS, 42 mass units more than **8**, corresponding to a molecular formula C₄₁H₅₂O₁₉. This evidence, combined with the close similarity of its ¹H and ¹³C NMR spectra (Table 3) to those of **8**, suggested that **9** is an acetylated derivative of **8**. The C-2 (δ_C 73.9) and H-30 (δ_H 4.21, s) signals of **8** were shifted to δ_C 82.1 and δ_H 4.64 in **9**, which revealed that **9** is a 2-*O*-acetyl derivative of **8**. The relative configuration was determined to be the same as that of **7** from the key NOESY crosspeaks.

Chuktabularins M (10) and N (11) showed molecular formulas of $C_{42}H_{54}O_{19}$ and $C_{43}H_{56}O_{19}$, as determined by the HRESIMS ions at *m*/*z* 885.3165 [M + Na]⁺ and 899.3309 [M + Na]⁺, respectively, which indicated the presence of one more CH₂ and C₂H₄ unit than in 9. The ¹H and ¹³C NMR spectra of 10 and 11 were similar to those of 9, including the characteristic signals of the ketal moiety, the propionyl at C-15, and the acetoxylated C-19 angular methyl group. The main differences were signals for an additional propionyl and isobutyryl group at C-12 in 10 and 11, instead of the acetyl in 9, which was deduced on the basis of their corresponding HMBC correlations. Thus, the structures of 10 and 11 were established as the 12-*O*-propionyl and the 12-*O*-isobutyryl derivative of 9, respectively.

The acetoxylated C-19 methylene exhibited two moderately broadened proton signals in the ¹H NMR spectra of **8**–11 recorded in CDCl₃ at room temperature (303 K). These compounds also exist as equilibrium mixtures of conformational isomers in solution, like compound **7**.

Chuktabularin O (12), a white, amorphous powder, was found to possess a molecular formula of $C_{39}H_{50}O_{18}$ from the HRESIMS ion at m/z 829.2901 [M + Na]⁺. The similarity of the ¹H and ¹³C NMR spectra of 12 when compared to those of 1 indicated that 12 possesses a 16-norphragmalin skeleton with a ketal moiety and a propionyl at C-15, which was confirmed by the observed HMBC correlations of the ketal resonance at δ_C 112.6 with the H-15a (δ_H 2.55, dd, J = 11.4, 7.8 Hz), H-30 (δ_H 4.57, s), and ethyl signals [δ_H 1.97 (q, J = 7.8 Hz, 2H); δ_H 1.08 (t, J = 7.8 Hz, 3H)]. Comparison of the 1D and 2D NMR data of 12 with those of 1



Figure 3. CD spectrum of 5a. Bold lines denote the electric transition dipole of the chromophores.



Figure 4. Key HMBC (\rightarrow) correlations of **7**.



Figure 5. Expanded C-19 methylene signals of **7** in variabletemperature ¹H NMR experiments.

revealed the absence of any oxygenated C-19 methylene proton signals; instead a normal C-19 methyl at $\delta_{\rm H}$ 1.54 (s, 3H) was present, correlating with C-5 ($\delta_{\rm C}$ 46.5), C-9 ($\delta_{\rm C}$ 76.4), and C-10 ($\delta_{\rm C}$ 54.2) in the HMBC spectrum (Figure 6). A singlet proton signal at $\delta_{\rm H}$ 4.45 was assignable to H-6 from the HMBC correlations observed with the quaternary carbon signals at $\delta_{\rm C}$ 46.5 (C-5) and 175.9 (C-7), which suggested that C-6 is oxygenated. Furthermore, on the basis of the chemical shifts of C-2 at $\delta_{\rm C}$ 82.3 and H-30 at $\delta_{\rm H}$ 4.57, along with the corresponding HMBC correlations, five acetyls were assignable to C-2 ($\delta_{\rm C}$ 82.3), C-3 ($\delta_{\rm C}$ 84.3), C-11 ($\delta_{\rm C}$ 72.7), C-12 ($\delta_{\rm C}$ 72.9), and C-17 ($\delta_{\rm C}$ 71.3), respectively. Strong cross-peaks in the NOESY experiment (Figure 6) indicated a β -orientation of H-5, H-11, H-12, H-17, H-30, and Me-28 and the α -orientation of H-3, H-14, Me-18, and H-29, as in compound 1. The NOESY correlation of H-5 with H-6 established H-6 as being in a β -orientation. Thus, the structure of **12** was established as shown.

Chuktabularin P (13), obtained as a white, amorphous powder, showed a molecular formula of $C_{41}H_{52}O_{19}$ as determined by the HRESIMS ion at m/z 871.3008 [M + Na]⁺. The ¹H and ¹³C NMR data (Table 4) showed close similarities to those of 12, except for the presence of an additional acetyl group. The differences identified were that the H-6 signal ($\delta_{\rm H}$ 4.45) of 12 was shifted downfield to $\delta_{\rm H}$ 5.56 in 13 and that the C-7 signal ($\delta_{\rm C}$ 175.9) of 12 shifted upfield to $\delta_{\rm C}$ 170.2 for 13 (Table 4). Hence, compound 13 could be assigned as the 6-*O*-acetyl derivative of 12, and this was supported by the HMBC correlations between H-6 and the acetyl carbonyl carbon ($\delta_{\rm C}$ 169.8). The relative configuration was determined to be the same as that of 12 using the NOESY spectrum. Thus, the structure of 13 was established as the 6-*O*-acetyl derivative of 12.

Chuktabularin Q (14) gave the molecular formula of $C_{37}H_{48}O_{16}$, as deduced from the HRESIMS ion at m/z 771.2837 [M + Na]⁺. The ¹H and ¹³C NMR data (Table 4) and the information obtained from the subsequent 2D NMR studies indicated that 14 possesses the 16-norphragmalin skeleton with a ketal moiety between the carbon skeleton and a propionyl at C-15, in the same manner as chuktabularin C.^{5d} Comparison of the ¹H and ¹³C NMR data of 14 and chuktabularin C revealed that the only differences were the chemical shifts of C-2 (δ_C 82.3) and H-30 (δ_H 4.65, s) in chuktabularin C, which were shifted upfield to δ_C 74.0 and δ_H 4.22 in 14, and indicated that chuktabularin Q is the 2-*O*-deacetyl derivative of chuktabularin C.^{7d} Thus, the structure of 14 was demonstrated as depicted, and the relative configuration was determined to be the same as chuktabularin C from the key NOESY correlations observed.

Chuktabularin S (15) was isolated as a white, amorphous powder, and its molecular formula was determined as $C_{36}H_{46}O_{16}$ from the HRESIMS ion at m/z 757.2686 [M + Na]⁺. The ¹H and ¹³C NMR data (Table 4) of 15 shared close similarities to those of 14, except for the presence of a methyl group signal in 15 in place of the ethyl signals observed for 14. In the HMBC spectrum of 15 (Figure 7), the observed correlations for the ketal resonance at δ_C 110.6 (C-1') with the H-15a (δ_H 2.53, dd, J = 11.4, 7.8 Hz), a methyl (δ_H 1.71), and H-30 (δ_H 4.20, s) signals suggested that 15 possesses an acetyl at C-15, forming a characteristic ketal moiety with the limonoid skeleton, in the same manner as chuktabularin A.^{7d} Furthermore, the four acetyl groups were assignable to C-3 (δ_C

Table 3. ¹H NMR and ¹³C NMR Data of Compounds 7–11 in CDCl₃

	7 ^a		8^{a}		9 ^{<i>a</i>}		10 ^{<i>a</i>}		11^b	
position	$\delta_{\rm H} (J \text{ in Hz})$	$\delta_{\rm C}$	δ_{H} (J in Hz)	$\delta_{\rm C}$	$\delta_{\mathrm{H}}~(J~\mathrm{in}~\mathrm{Hz})$	$\delta_{\rm C}$	δ_{H} (J in Hz)	$\delta_{\rm C}$	δ_{H} (J in Hz)	$\delta_{\rm C}$
1		84.8		85.1		84.7		84.8		84.8
2		82.0		73.9		82.1		82.1		82.2
3	5.24, s	83.0	5.14, s	85.5	5.24, s	83.1	5.24, s	83.1	5.23, s	83.1
4		45.3		44.6		45.5		45.4		45.5
5	2.70, brd (12.0)	41.6	2.70, brd (12.6)	42.1	2.70, brd (12.0)	41.6	2.70, brd (12.0)	41.6	2.69, brd (12.2)	41.6
6a	2.40, m ^c	33.2	2.41, m ^c	33.2	2.42, m ^c	33.2	2.40, m ^c	33.3	2.36, m ^c	33.4
6b 7	2.22, m ^e	172.1	2.20, dd (16.5, 12.0)	172.0	2.20, m ^e	172.1	2.20, m ^e	172.1	2.20, m ^e	172.0
0		1/3.1		1/3.2		1/3.1		1/3.1		1/3.0
0		09.9 77 0		90.0 77 1		09.0 77.0		09.0 77.0		09.0 77.0
10		53.8		53.6		53.8		53.8		53.8
11	570 d (36)	72.0	567 d (36)	72.0	571 d (36)	72.0	5.72 brs	72.0	5.72 brs	72.2
12	5.66. d (3.6)	72.8	5.66. d (3.6)	72.6	5.67. d (3.6)	72.8	5.71, brs	72.5	5.72, brs	72.1
13		41.6		41.6		41.5		41.5		41.6
14	3.19, dd (12.0, 7.8)	44.4	3.17, dd (11.4, 8.4)	43.9	3.13, dd (12.0, 7.8)	44.0	3.17, dd (12.0, 7.8)	44.0	3.22, dd (12.0, 8.2)	44.2
15a	2.55, dd (11.4, 7.8)	35.4	2.57, dd (12.0, 8.4)	33.3	2.55, dd (11.4, 7.8)	34.0	2.55, dd (11.4, 7.8)	34.0	2.55, m ^c	34.1
15b	1.92, dd (12.0, 11.4)		1.95, dd (12.0, 11.4)		1.88, dd (12.0, 11.4)		1.88, dd (12.0, 11.4)		1.88, dd (12.0, 11.7)	
17	6.15, s	71.2	6.24, s	71.0	6.16, s	71.3	6.16, s	71.3	6.16, s	71.4
18	0.92, s, 3H	19.0	0.95, s, 3H	18.6	0.92, s, 3H	19.0	0.91, s, 3H	19.0	0.90, s, 3H	19.1
19a	4.64, brd (11.7)	66-67	4.67, brd (12.6)	66.8	4.63, brd (11.7)	66-67	4.66, brd (11.7)	66-67	4.67, brd (11.7)	66-67
19b	4.46, brd (11.7)		4.35, brd (12.6)		4.47, brd (11.7)		4.46, brd (11.7)		4.43, brd (11.7)	
20		122.3		122.1		122.3		122.4	T (0.1	122.4
21	7.64, brs	140.4	7.66, brs	140.7	7.64, brs	140.4	7.64, brs	140.4	7.63, brs	140.4
22	6.50, d(0.6)	109.6	0.55, brs 7.20 , 1.11 , (1.8)	109.6	(0.51, 0, (0.6))	109.6	6.52, d(1.2)	109.6	6.52, d(1.2)	109.6
23	7.56, t-like (1.6)	145.1	7.39, t-like (1.0)	145.5	7.30, t-like(1.2)	145.1	7.50, t-like (1.0)	145.1	7.56, t-like(1.5)	145.1
20	2.30, 8, 511	40.2	2.05°	30.7	2.10°	10.1	2.00°	40.2	2.10°	10.2
29h	1.85 d (11.4)	40.2	1.93 ^c	39.1	1.86 ^c	40.2	2.09 1.87 d (11.4)	40.2	2.12 1.87 ^c	40.5
30	4.64. s	70.8	4.21. s	71.0	4.64. s	70.5	4.64. s	70.4	4.66. s	70.5
1'		110.5	, .	113.2		112.8		112.8	, .	112.8
2'	1.63, s, 3H	18.7	2.04, q, 2H (7.8)	26.1	1.97, q, 2H (7.8)	26.0	1.96, q, 2H (7.8)	26.0	1.96, q, 2H (7.5)	26.0
3'			1.09, t, 3H (7.8)	8.2	1.07, t, 3H (7.8)	7.9	1.07, t, 3H (7.8)	7.9	1.07, t, 3H (7.5)	7.9
OH-1	4.82, s		4.42, s		4.83, s		4.84, s		4.82, s	
OH-9	3.55, s		3.53, s		3.54, s		3.57, s		3.58, s	
OMe-7	3.63, s, 3H	51.9	3.64, s, 3H	51.9	3.63, s, 3H	51.9	3.63, s, 3H	51.8	3.63, s, 3H	51.8
OAc-2		169.7				169.5		169.5		169.4
<u></u>	2.10, s, 3H	21.5		1.00.0	2.10, s, 3H	21.5	2.09, s, 3H	21.5	2.07, s, 3H	21.4
OAc-3	2.40 - 211	169.4	2.42 - 211	169.9	2.50 . 211	169.4	2.50 . 211	169.4	2.40 - 211	169.3
OAc 11	2.49, 8, 5H	21.1 160.3	2.43, 8, 3H	21.3 160.3	2.30, s, 3H	21.1 160.3	2.30, s, 3H	21.1	2.49, 8, 3H	21.1 160.0
OAC-11	105 c 3H	20.0	108 c 3H	20.0	105 s 3H	20.0	105 s 3H	20.0	104 c 3H	20.8
OAc_{-12}	1.95, 8, 511	169.5	1.90, 3, 511	169.5	1.95, 5, 511	169.4	1.95, 5, 511	20.9	1.94, 5, 511	20.8
0/10-12	2.06 s 3H	20.5	2.07 s 3H	20.5	2.07 s 3H	20.5				
OCOCH ₂ CH ₃ -12	, .,		, .,		, .,			173.0		
2 - 5							2.34, q, 2H (7.8)	27.4		
							1.15, t, 3H (7.8)	9.3		
OCOCH(CH ₃) ₂ -12										175.6
									2.55, m	34.1
									1.18, d, 3H (6.9)	19.1
									1.15, d, 3H (6.9)	18.9
OAc-17	2.10 . 211	168.8	0.10 . 011	169.0	2.10 . 211	168.8	2.10 . 211	168.8	2.00 . 211	168.7
OA a 10	2.10, s, 3H	20.5	2.13, 8, 3H	20.8	2.10, s, 3H	20.5 160 5	2.10, s, 3H	20.5	2.08, s, 3H	20.6
0AC-19	207 s 3H	20.0	2 10 s 3H	109./	208 s 3H	20.0	2 08 s 3H	21.0	200 s 3H	21.0
-	2.07, 8, 517	20.9	2.10, 8, 317	41.J	2.00, 8, 311	20.9	2.00, 8, 317	21.0	2.07, 8, 317	21.0

^a Recorded at 600 MHz (¹H) and 150 MHz (¹³C). ^b Recorded at 500 MHz (¹H) and 125 MHz (¹³C). ^c Signal pattern unclear due to overlapping.



Figure 6. Selected HMBC (\rightarrow) and NOE (\leftrightarrow) correlations of 12.

Table 4. ¹H NMR and ¹³C NMR Data of **12–16** (in CDCl₃)

	12^a		13 ^a		14^b		15^a		16 ^{<i>a</i>}	
position	$\delta_{\rm H} (J \text{ in Hz})$	$\delta_{\rm C}$	$\delta_{\rm H} (J \text{ in Hz})$	$\delta_{\rm C}$	$\delta_{\rm H} (J \text{ in Hz})$	$\delta_{\rm C}$	$\delta_{\rm H} (J \text{ in Hz})$	$\delta_{\rm C}$	$\delta_{\rm H} (J \text{ in Hz})$	$\delta_{\rm C}$
1		84.4		84.4		85.1		85.1		84.5
2		82.3		82.2		74.0		73.8		82.1
3	5.14, s	84.3	5.19, s	83.8	5.14, s	85.9	5.14, s	85.8	5.22, s	83.3
4		45.2		45.1		44.7		44.6		45.4
5	2.60, brs	46.5	2.84, brs	45.4	2.60, brd (11.8)	41.7	2.58, brd (11.8)	41.7	2.60, brd (11.8)	41.2
6a	4.45, brs	71.1	5.56, d (0.6)	71.2	2.43, brd (16.5)	34.2	2.44, brd (16.2)	34.1	2.52, brd (16.2)	34.1
6b					2.24, dd (16.5, 12.2)		2.25, brd (16.2, 12.0)		2.24, dd (16.2, 12.0)	
7		175.9		170.2		173.3		173.3		173.3
8		90.0		90.0		90.2		90.3		90.2
9		76.4		76.4		76.2		76.1		78.0
10		54.2		54.2		52.2		52.0		52.3
11	5.63, d (3.6)	72.7	5.63, d (3.0)	72.6	5.65, d (2.3)	72.5	5.66, d (3.3)	72.4	5.51, d (3.0)	73.4
12	5.68, d (3.6)	72.9	5.73, d (3.0)	73.0	5.64, d (2.3)	73.1	5.63, d (3.3)	73.0	4.16, d (3.0)	72.8
13		41.0		41.6		41.7		41.7		42.5
14	3.11, dd (12.0, 7.8)	44.4	3.09, dd (12.0, 7.8)	44.4	3.06, dd (11.8, 8.1)	44.7	3.09, dd (12.0, 7.8)	44.8	3.00, dd (12.0, 7.8)	43.8
15a	2.55, dd (11.4, 7.8)	34.0	2.55, dd (11.4, 7.8)	34.0	2.51, dd (11.8, 8.1)	33.5	2.53, dd (11.4, 7.8)	35.2	2.54, dd (11.4, 7.8)	35.5
15b	1.87, dd (12.0, 11.4)		1.87, dd (12.0, 11.4)		1.91, brd (12.0)		1.97, brd (12.0)		1.91, brd (12.0)	
17	6.11, s	71.3	6.12, s	71.2	6.24, s	71.1	6.24, s	71.0	6.09, s	71.5
18	0.90, s, 3H	19.2	0.91, s, 3H	19.2	0.95, s, 3H	18.7	0.95, s, 3H	19.1	1.05, s, 3H	19.1
19	1.54, s, 3H	17.7	1.26, s, 3H	17.0	1.19, s, 3H	17.7	1.19, s, 3H	17.7	1.21, s, 3H	17.3
20		122.5		122.3		122.1		122.0		122.6
21	7.63, brs	140.3	7.66, brs	140.4	7.67, brs	140.7	7.68, brs	140.7	7.64, brs	140.6
22	6.49, d (1.2)	109.6	6.51, d (1.2)	109.5	6.54, brs	109.6	6.54, brs	109.6	6.47, brs	110.0
23	7.39, t-like (1.8)	143.1	7.39, t-like (1.8)	143.2	7.38, brs	143.2	7.39, brs	143.3	7.36, brs	142.8
28	0.85, s, 3H	16.7	0.91, s, 3H	16.6	0.80, s, 3H	16.1	0.79, s, 3H	16.0	0.77, s, 3H	16.0
29a	2.45, d (11.4)	41.5	2.21, dd (11.4, 1.2)	41.4	1.88, d (11.2)	39.6	1.89, d (11.4)	39.6	1.83, d (11.4)	40.0
296	1.76, dd (11.4, 1.8)	70 7	1.84, d (11.4)	70 7	1.72, d (11.2)	71.7	1./4, d (11.4)	72.0	1.80, d (11.4)	71.1
30	4.57, s	/0./	4.60, s	/0./	4.22, s	/1./	4.20, s	110.6	4.22, s	/1.1
1	1.07	112.0	1.00	112.8	2.02 - 211.(7.5)	112.1	1.71 . 211	110.0	1.71 . 211	110.7
2	1.97, q, 2H, (7.8)	26.0	1.98, q, 2H, (7.8)	26.0	2.02, q, 2H, (7.5)	26.2	1./1, s, 3H	18.7	1./1, S, 3H	18.9
3 ОН 1	1.08, t, 3H, (7.8)	7.9	1.07, t, 3H, (7.8)	7.9	1.09, t, 3H, (7.5)	8.2	4.27		4.60 a	
011-1	4.70, 8		4.70, 8		4.22, 8		4.27, 8		4.00, 8	
OH-2 OMe 7	3 77 6 31	52.5	2.68 ° 311	52.6	2.62 . 211	517	3.01, 8 3.64 s 31	51.8	2.64 s 2U	51.8
OAc^2	5.77, 8, 511	160.8	5.00, 8, 511	160.7	5.05, 8, 511	51.7	5.04, 8, 511	51.0	5.04, 8, 511	160.7
OAC-2	2.07 8.21	21.0	2 10 6 31	21.0					2.08 5.21	21.0
$0 \wedge a^3$	2.07, 8, 511	160.2	2.10, 8, 511	160.2		160.0		170.0	2.00, 8, 511	160.5
OAC-3	2.45 s 3H	21.0	246 s 3H	21.0	2.41 s 3H	21.3	2 4 2 s 3H	21.3	2.46 s 3H	21.1
OAc-6	2.45, 5, 511	21.0	2.40, 5, 511	169.8	2.41, 5, 511	21.5	2.42, 3, 511	21.5	2.40, 5, 511	21.1
0/10-0			2.20 s 3H	21.2						
OAc-11		169.6	2.20, 3, 511	169.8		169.4		169 5		169.9
0/10/11	194 s 3H	21.4	2.02 s 3H	21.5	194 s 3H	21.0	195 s 3H	21.0	2.08 s 3H	21.4
OAc-12		169.4	, 0, 011	169.2		169.2		169.2	2.00, 0, 011	21.7
0/10/12	2.10. s. 3H	20.7	2.10, s. 3H	20.6	2.08. s. 3H	20.6	2.10. s. 3H	20.6		
OAc-17	, 0, 011	168.9	, 0, 011	168.8	, 0, 011	168.8	, 0, 011	169.0		169.0
	2.10, s. 3H	20.5	2.10, s, 3H	20.5	2.12, s. 3H	20.8	2.14, s. 3H	20.9	2.09. s. 3H	20.6
	, -, -		, -, -				, -, -			

^a Recorded at 600 MHz (¹H) and 150 MHz (¹³C). ^b Recorded at 500 MHz (¹H) and 125 MHz (¹³C).





85.8), C-11 ($\delta_{\rm C}$ 72.4), C-12 ($\delta_{\rm C}$ 73.0), and C-17 ($\delta_{\rm C}$ 71.0), on the basis of their corresponding HMBC correlations. Thus, the structure of **15** was demonstrated as the 2-*O*-deacetyl derivative of chuktabularin A, and the relative configurations of both compound were determined to be the same from the NOESY spectra.

Chuktabularin T (16), a white, amorphous powder, gave a molecular formula of $C_{36}H_{46}O_{16}$ as established by the HRESIMS

ion at m/z 757.2681 [M + Na]⁺. This evidence, when combined with the ¹H and ¹³C NMR spectroscopic data (Table 4) and information obtained from subsequent 2D NMR studies, indicated that **16** is a positional isomer of **15**. The H-12 signal at $\delta_{\rm H}$ 5.64 in **15** appeared at $\delta_{\rm H}$ 4.16 in **16** and the C-2 signal at $\delta_{\rm C}$ 73.8 was shifted downfield to $\delta_{\rm C}$ 82.1, consistent with the transfer of an acetyl from C-12 in **15** to C-2 in **16**. This matched the absence of HMBC correlations between H-12 and the acetyl carbon. Thus, the structure of **16** was determined as the 12-*O*-deacetyl derivative of chuktabularin A.

Four known compounds were identified as chuktabularins A-D by comparison of their physical and spectral data (IR, ESIMS, and NMR) with reported values.^{7d} By correlation with chuktabularin I (5) on the biosynthetic origin and structure, the absolute configuration of the key chiral carbons in compounds 1-4 and 6-16 could be proposed as depicted.

Experimental Section

General Experimental Procedures. Optical rotations were measured with a JASCO P-1020 polarimeter. CD spectra were obtained on a JASCO 810 spectropolarimeter. IR (KBr disks) spectra were recorded on a Bruker Tensor 27 spectrometer. NMR spectra were recorded on Bruker ACF-500 and -600 NMR instruments (¹H: 500 or 600 MHz, ¹³C: 125 or 150 MHz), with TMS as internal standard. Mass spectra were obtained on a MS Agilent 1100 Series LC/MSD ion-trap mass

spectrometer (ESIMS) and a Mariner ESITOF spectrometer (HRES-IMS), respectively. All solvents used were of analytical grade (Jiangsu Hanbang Science and Technology. Co., Ltd.). Silica gel (Qingdao Haiyang Chemical Co., Ltd.), Sephadex LH-20 (Pharmacia), and RP-C₁₈ (40–63 μ m, Fuji) were used for column chromatography. Preparative HPLC was carried out using an Agilent 1100 Series instrument with a Shim-park RP-C₁₈ column (20 × 200 mm) and a 1100 Series multiple wavelength detector.

Plant Material. The air-dried stem bark of *C. tabularis* var. *velutina* was collected from Xishuangbanna, Yunnan Province, People's Republic of China, in March 2007, and was authenticated by Professor Mian Zhang of the Research Department of Pharmacognosy, China Pharmaceutical University. A voucher specimen (no. 2006-MML) has been deposited in the Department of Natural Medicinal Chemistry, China Pharmaceutical University.

Extraction and Isolation. The air-dried stem bark (10 kg) was extracted by refluxing with 95% ethanol three times. The EtOH extract was concentrated under reduced pressure (2000 g) and then extracted with CHCl₃ to give a chloroform extract (300 g). The oily chloroform extract was dissolved in 2 L of 50% MeOH and H₂O and then extracted with petroleum ether. After removal of the fatty components, 210 g of extract was obtained, which was subjected to passage over a silica gel column eluted with CHCl3-MeOH in a gradient (1:0 to 1:2), to afford eight fractions (Fr. A-H), monitored by TLC. Fr. C (22 g) was chromatographed on a column of silica gel, eluted successively with a gradient of petroleum ether-EtOAc (4:1 to 1:2), to give eight subfractions (Fr. C1-C8). Fr. C5 was chromatographed on a column of reversed-phase C₁₈ silica gel, eluted with MeOH-H₂O (5:5 to 7:3), to give three subfractions (Fr. C5a-C5e). Fr. C5c was separated by preparative HPLC using CH₃OH-H₂O (64:35, 10 mL/min) as the mobile phase to give 11 (6 mg). Fr. C7 was chromatographed on a column of reversed-phase C₁₈ silica gel, eluted with MeOH-H₂O (5:5 to 7:3), to give three subfractions (Fr. C7a-C7e). Fr. C7b was separated by preparative HPLC using CH₃OH-H₂O (60:40, 10 mL/min) as the mobile phase to give 9 (20 mg), 13 (6 mg), and a mixture that contained 8, and then the mixture was purified using CH₃CN-H₂O (50:50, 10 mL/ min) as the mobile phase to give 8 (12 mg). Fr. C7c was separated by preparative HPLC using CH₃CN-H₂O (55:45, 10 mL/min) as the mobile phase to give 2 (10 mg) and a mixture that contained 10. This mixture was purified using CH_3OH-H_2O (65:35, 10 mL/ min) as the mobile phase to give 10 (7 mg). Fr. C7e was separated by preparative HPLC with CH₃OH-H₂O (65:35, 10 mL/min) as the mobile phase to give 3 (8 mg). Fr. C8 was chromatographed on a column of reversed-phase C₁₈ silica gel eluted with MeOH-H₂O (5:5 to 7:3) to give three subfractions (Fr. C8a-C8f). Fr. C8c was separated by preparative HPLC using CH₃OH-H₂O (55:45, 10 mL/ min) as the mobile phase to give 7 (26 mg), chuktabularin A (30 mg), and chuktabularin B (26 mg). Fr. C8d was separated by preparative HPLC using CH₃OH-H₂O (60:40, 10 mL/min) as the mobile phase to afford chuktabularin C (15 mg). Fr. C8e was separated by preparative HPLC using CH₃OH-H₂O (60:40, 10 mL/ min) as the mobile phase to yield 14 (10 mg), chuktabularin C (18 mg), and a mixture that contained 1. This mixture was purified by preparative HPLC, with CH₃CN-H₂O (45:55, 10 mL/min) as the mobile phase to give 1 (8 mg). Fr. D (30 g) was chromatographed on a column of silica gel eluted successively with a gradient of petroleum ether-EtOAc (5:2 to 1:2) to give seven subfractions (Fr. D1-D7). Fr. D4 was chromatographed on a column of reversedphase C₁₈ silica gel eluted with MeOH-H₂O (5:5 to 7:3) to give four subfractions (Fr. D4a-D4d). Fr. D4b was separated by preparative HPLC using CH₃CN-H₂O (42:58, 10 mL/min) as the mobile phase to give 4 (20 mg). Fr. D4c was purified by preparative HPLC using CH₃CN-H₂O (45:55, 10 mL/min) as the mobile phase to give 16 (22 mg). Fr. D5 was chromatographed on a column of reversed-phase C₁₈ silica gel eluted with MeOH-H₂O (5:5 to 7:3) to give four subfractions (Fr. D5a-D5d). Fr. D5b was separated by preparative HPLC using CH₃OH-H₂O (55:45, 10 mL/min) as the mobile phase to give a mixture containing 15. This mixture was purified by preparative HPLC using CH₃CN-H₂O (42:58, 10 mL/ min) as the mobile phase to give 15 (8 mg). Fr. D5c was separated by preparative HPLC using CH₃OH-H₂O (55:45, 10 mL/min) as the mobile phase to afford 12 (6 mg). Fr. E (20 g) was chromatographed on a column of reversed-phase C₁₈ silica gel, eluted with MeOH-H₂O (4:6 to 7:3), to give six subfractions (Fr. E1-E6). Fr. E5 was chromatographed on a column of reversed-phase C_{18} silica gel, eluted with MeOH-H₂O (4:6 to 7:3), to give four subfractions (Fr. E5a-E5d). Fr. E5b was separated by preparative HPLC using CH₃CN-H₂O (40:60, 10 mL/min) as the mobile phase to give **5** (100 mg) and **6** (10 mg).

Chuktabularin E (1): colorless crystals (MeOH-H₂O); $[\alpha]^{25}_{D}$ +28 (*c* 0.075, CH₃OH); IR (KBr) ν_{max} 3418, 2980, 1755, 1636, 1374, 1211, 1031, 916, 601 cm⁻¹; ¹H NMR, see Table 1; ¹³C NMR, see Table 2; negative ESIMS *m*/*z* 791.1 [M + CH₃COO]⁻ (100); positive ESIMS *m*/*z* 750.2 [M + NH₄]⁺ (100); HRESIMS *m*/*z* 755.2527 [M + Na]⁺ (calcd C₃₆H₄₄O₁₆Na, 755.2522).

Chuktabularin F (2): white, amorphous powder; $[\alpha]^{25}_{D} + 24$ (*c* 0.095, CH₃OH); IR (KBr) ν_{max} 3424, 2984, 1751, 1639, 1375, 1219, 1041, 926, 600 cm⁻¹; ¹H NMR, see Table 1; ¹³C NMR, see Table 2; negative ESIMS *m/z* 823.4 [M + Cl]⁻ (100); positive ESIMS *m/z* 806.3 [M + NH₄]⁺ (100); HRESIMS *m/z* 811.2794 [M + Na]⁺ (calcd C₃₉H₄₈O₁₇Na, 811.2784).

Chuktabularin G (3): white, amorphous powder; $[\alpha]^{25}_{D} + 23$ (*c* 0.080, CH₃OH); IR (KBr) ν_{max} 3432, 2981, 1753, 1638, 1375, 1219, 1040, 924, 602 cm⁻¹; ¹H NMR, see Table 1; ¹³C NMR, see Table 2; negative ESIMS *m/z* 837.5 [M + Cl]⁻ (100); positive ESIMS *m/z* 820.2 [M + NH₄]⁺ (100); HRESIMS *m/z* 825.2938 [M + Na]⁺ (calcd C₄₀H₅₀O₁₇Na, 825.2940).

Chuktabularin H (4): white, amorphous powder; $[\alpha]^{25}_{D} + 17$ (*c* 0.075, CH₃OH); IR (KBr) ν_{max} 3440, 2980, 1750, 1637, 1374, 1234, 1025, 909, 603 cm⁻¹; ¹H NMR, see Table 1; ¹³C NMR, see Table 2; negative ESIMS *m/z* 689.2 [M - H]⁻ (100); positive ESIMS *m/z* 708.2 [M + NH₄]⁺ (100); HRESIMS *m/z* 713.2412 [M + Na]⁺ (calcd C₃₄H₄₂O₁₅Na, 713.2416).

Chuktabularin I (5): white, amorphous powder; $[\alpha]^{25}_{D} + 25$ (*c* 0.090, CH₃OH); IR (KBr) ν_{max} 3442, 2981, 1752, 1642, 1374, 1242, 1037, 902, 601 cm⁻¹; ¹H NMR, see Table 1; ¹³C NMR, see Table 2; negative ESIMS *m/z* 689.3 [M - H]⁻ (100); positive ESIMS *m/z* 708.4 [M + NH₄]⁺ (100); HRESIMS *m/z* 713.2419 [M + Na]⁺ (calcd C₃₄H₄₂O₁₅Na, 713.2416).

Chuktabularin J (6): white, amorphous powder; $[\alpha]^{25}_{D} + 19$ (*c* 0.070, CH₃OH); IR (KBr) ν_{max} 3432, 2979, 1738, 1639, 1374, 1234, 1030, 899, 602 cm⁻¹; ¹H NMR, see Table 1; ¹³C NMR, see Table 2; negative ESIMS *m/z* 683.4 [M + Cl]⁻ (100); positive ESIMS *m/z* 666.2 [M + NH₄]⁺ (100); HRESIMS *m/z* 671.2331 [M + Na]⁺ (calcd C₃₂H₄₀O₁₄Na, 671.2310).

Chuktabularin K (7): white, amorphous powder; $[\alpha]_{25}^{25} + 23$ (*c* 0.090, CH₃OH); IR (KBr) ν_{max} 3436, 2986, 1751, 1641, 1372, 1239, 1039, 876, 602 cm⁻¹; ¹H and ¹³C NMR, see Table 3; negative ESIMS *m*/*z* 893.3 [M + CH₃COO]⁻ (100); positive ESIMS *m*/*z* 852.2 [M + NH₄]⁺ (100); HRESIMS *m*/*z* 857.2841 [M + Na]⁺ (calcd C₄₀H₅₀O₁₉Na, 857.2839).

Chuktabularin L (8): white, amorphous powder; $[\alpha]^{25}_{D} + 28$ (*c* 0.075, CH₃OH); IR (KBr) ν_{max} 3443, 2980, 1749, 1637, 1374, 1224, 1031, 898, 603 cm⁻¹; ¹H and ¹³C NMR, see Table 3; negative ESIMS *m*/*z* 841.6 [M + Cl]⁻ (100); positive ESIMS *m*/*z* 829.2 [M + Na]⁺ (100); HRESIMS *m*/*z* 829.2891 [M + Na]⁺ (calcd C₃₉H₅₀O₁₈Na, 829.2889).

Chuktabularin M (9): white, amorphous powder; $[\alpha]^{25}_{D} + 29$ (*c* 0.085, CH₃OH); IR (KBr) ν_{max} 3437, 2983, 1751, 1640, 1373, 1238, 1038, 903, 602 cm⁻¹; ¹H and ¹³C NMR, see Table 3; negative ESIMS *m*/*z* 805.3 [M - COCH₃]⁻ (100); positive ESIMS *m*/*z* 866.4 [M + NH₄]⁺ (100); HRESIMS *m*/*z* 871.3001 [M + Na]⁺ (calcd C₄₁H₅₂O₁₉Na, 871.2995).

Chuktabularin N (10): white, amorphous powder; $[\alpha]_{D}^{25} + 22$ (*c* 0.075, CH₃OH); IR (KBr) ν_{max} 3435, 2983, 1751, 1639, 1372, 1221, 1038, 903, 602 cm⁻¹; ¹H and ¹³C NMR, see Table 3; negative ESIMS *m*/*z* 897.5 [M + Cl]⁻ (100); positive ESIMS *m*/*z* 880.4 [M + NH₄]⁺ (100); HRESIMS *m*/*z* 885.3165 [M + Na]⁺ (calcd C₄₂H₅₄O₁₉Na, 885.3152).

Chuktabularin O (11): white, amorphous powder; $[\alpha]_{D}^{25} + 29$ (*c* 0.050, CH₃OH); IR (KBr) ν_{max} 3432, 2980, 1751, 1636, 1372, 1221, 1038, 903, 602 cm⁻¹; ¹H and ¹³C NMR, see Table 3; negative ESIMS *m*/*z* 911.7 [M + Cl]⁻ (100); positive ESIMS *m*/*z* 894.3 [M + NH₄]⁺ (100); HRESIMS *m*/*z* 899.3309 [M + Na]⁺ (calcd C₄₃H₅₆O₁₉Na, 899.3308).

Chuktabularin P (12): white, amorphous powder; $[\alpha]_{D}^{25} + 30$ (*c* 0.070, CH₃OH); IR (KBr) ν_{max} 3452, 2982, 1753, 1640, 1373, 1243, 1035, 902, 602 cm⁻¹; ¹H and ¹³C NMR, see Table 4; negative ESIMS

m/z 805.1 [M – H]⁻ (100); positive ESIMS m/z 824.3 [M + NH₄]⁺ (100); HRESIMS m/z 829.2901 [M + Na]⁺ (calcd C₃₉H₅₀O₁₈Na, 829.2889).

Chuktabularin Q (13): white, amorphous powder; $[\alpha]^{25}_{D} + 29$ (*c* 0.075, CH₃OH); IR (KBr) ν_{max} 3446, 2983, 1758, 1638, 1373, 1219, 1037, 902, 602 cm⁻¹; ¹H and ¹³C NMR, see Table 4; negative ESIMS *m*/*z* 805.3 [M - COCH₃]⁻ (100); positive ESIMS *m*/*z* 866.3 [M + NH₄]⁺ (100); HRESIMS *m*/*z* 871.3008 [M + Na]⁺ (calcd C₄₁H₅₂O₁₉Na, 871.2995).

Chuktabularin R (14): white, amorphous powder; $[\alpha]^{25}_{D} + 8$ (*c* 0.050, CH₃OH); IR (KBr) ν_{max} 3456, 2979, 1747, 1638, 1374, 1225, 1030, 898, 603 cm⁻¹; ¹H and ¹³C NMR, see Table 4; negative ESIMS *m*/*z* 807.1 [M + CH₃COO]⁻ (100); positive ESIMS *m*/*z* 766.2 [M + NH₄]⁺ (100); HRESIMS *m*/*z* 771.2837 [M + Na]⁺ (calcd C₃₇H₄₈O₁₆Na, 771.2835).

Chuktabularin S (15): white, amorphous powder; $[\alpha]^{25}_{D} + 26$ (*c* 0.050, CH₃OH); IR (KBr) ν_{max} 3454, 2981, 1752, 1638, 1373, 1226, 1030, 875, 603 cm⁻¹; ¹H and ¹³C NMR, see Table 4; negative ESIMS *m*/*z* 779.2 [M + CH₃COO]⁻ (100); positive ESIMS *m*/*z* 752.1 [M + NH₄]⁺ (100); HRESIMS *m*/*z* 757.2686 [M + Na]⁺ (calcd C₃₆H₄₆O₁₆Na, 757.2678).

Chuktabularin T (16): white, amorphous powder; $[\alpha]^{25}_{D}$ +4 (*c* 0.080, CH₃OH); IR (KBr) ν_{max} 3448, 2986, 1745, 1640, 1373, 1227, 1034, 876, 603 cm⁻¹; ¹H and ¹³C NMR, see Table 4; negative ESIMS *m*/*z* 769.5 [M + Cl]⁻ (100); positive ESIMS *m*/*z* 752.3 [M + NH₄]⁺ (100); HRESIMS *m*/*z* 757.2681 [M + Na]⁺ (calcd C₃₆H₄₆O₁₆Na, 757.2678).

X-ray crystallographic data for 1: $C_{36}H_{44}O_{16}$, M = 732.71, orthorhombic, dimensions $0.23 \times 0.20 \times 0.18$ mm, d = 1.400 g/cm³, space group $P2_12_12_1$, Z = 4, a = 10.4375(14) Å, b = 16.4155(18) Å, c = 20.288(2) Å, $\alpha = \beta = \gamma = 90^{\circ}$, V = 3476.1(7) Å³, reflections collected/unique 17 981/6140 ($R_{int} = 0.0543$), number of observation [$I > 2\sigma(I)$] 6140, parameters 482, final R indices [$I > 2\sigma(I)$]: $R_1 = 0.0594$, $wR_2 = 0.1242$.

Colorless crystals of **1** was obtained from the mixture MeOH $-H_2O$. Crystal data were obtained on a Bruker Smart-1000 CCD with a graphite monochromator, with Mo K α radiation ($\lambda = 0.71073$ Å) at 298(2) K. The crystal structure was solved by direct methods using SHELX-97¹² and expanded using difference Fourier techniques, refined by SHELX-97.¹³ Crystallographic data for the structure of **1** have been deposited in the Cambridge Crystallographic Data Centre with a deposition number of CCDC 743107. Copies of these data can be obtained free of charge via the Internet at www.ccdc.cam.ac.uk/conts/retrieving.html or from the Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB21EZ, UK [fax (+44) 1223 336 033; or e-mail: deposit@ccdc.cam.ac.uk].

Preparation of the *p*-Chlorobenzoate of Chuktabularin I (5). Compound 5 (22 mg) was treated with p-chlorobenzoic acid (50 mg), DCC (50 mg), and a catalytic amount of DMAP in CHCl₃ (4 mL), at room temperature for 48 h. The reaction mixture was purified by silica gel column chromatography (eluted with CHCl3 and CH3OH) and passage over Sephadex LH-20 using CH₂Cl₂-CH₃OH (1:1) as the mobile phase to give the 11,12-di-p-chlorobenzoate (5a, 10 mg): white amorphous powder; CD (CH₃OH, $\Delta \epsilon$) 212 (-1.17), 232 (+1.32), 255 (-4.30) nm; ¹H NMR (CDCl₃, 300 MHz) δ 7.55 (1H, s, H-21), 7.39 (1H, s, H-23), 6.45 (1H, brs, H-22), 6.24 (1H, s, H-17), 6.05 (1H, d, J = 3.3 Hz, H-11), 5.80 (1H, d, J = 3.3 Hz, H-12), 5.40 (1H, s, H-3), 5.02 (1H, d, J = 12.6 Hz, H-19a), 4.77 (1H, s, H-3), 4.02 (1H, d, J = 12.6, H-19b), 3.50 (1H, dd, J = 11.7, 7.9, H-14), 2.62 (3H, s, OAc-3), 2.13 (6H, s, OAc-2, OAc-17), 1.96-2.68 (8H, m, H-6, H-15, H-29, H-2'), 1.12 (3H, t, J = 7.5 Hz, H-3'), 0.96 (3H, s, H-18), 0.89 (3H, s, H-28); ¹³C NMR (CDCl₃, 125 MHz) δ 171.0 (C-7), 143.4 (C-23), 140.3 (C-21), 122.4 (C-20), 113.5 (C-1'), 109.6 (C-22), 89.6 (C-8), 85.6 (C-1), 83.1 (C-3), 81.3 (C-2), 75.9 (C-9), 72.6 (C-11), 72.6 (C-12), 71.3 (C-17), 70.9 (C-30), 69.2 (C-19), 52.4 (C-10), 45.3 (C-4), 44.2 (C-14), 41.8 (C-13), 40.2 (C-5), 38.9 (C-29), 34.1 (C-15), 30.9 (C-6), 26.0 (C-2'), 19.3 (C-18), 15.0 (C-28), 7.8 (C-3'), [169.5, 169.1, 168.7, 21.1, 20.8, 20.5] (OAc-2, OAc-3, OAc-17), [165.1, 164.2, 140.9, 140.6, 132.0, 131.1, 129.3, 128.5, 127.0, 126.1] (p-chlorobenzoyl); negative ESIMS m/z 1001.8 [M + Cl]⁻ (100); positive ESIMS m/z 984.8 [M + NH₄]⁺ (100).

Acknowledgment. This research work was supported by the Key Project of National Natural Science Foundation of China (Grant No. 30830116).

Supporting Information Available: Biogenetic scheme for compounds 1–16; enlarged X-ray structure of 1; ¹H NMR, ¹³C NMR, ESIMS, and HRESIMS spectra of 1–16. This material is available free of charge via the Internet at http://pubs.acs.org.

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NP900734C